TOXICITY AND CHARACTERISTIC LEACHING PROCEDURE  EPA 1311 1992  Page 1 of 3						
Facility Name:	VELAP ID					
Assessor Name:Analyst Name:	Inspection Date					
Relevant Aspect of Standards	Method Reference	Y	N	N/A	Comments	
Records Examined: SOP Number/ Revision/ Date		Analyst:				
Sample ID: Date of Sample Prepar	ation:	Date of Analysis:				
Were all reagents used reagent grade?	5.1					
Was Extraction Fluid #1 prepared by bringing 5.7 mL of glacial acetic acid and 64.3 mL of 1N NaOH to 1 Liter with reagent water?	5.7.1					
Was Extraction Fluid #2 prepared by bringing 5.7 mL of glacial acetic acid to 1 Liter with reagent water?	5.7.2					
Was the pH of Extraction Fluid #1 verified to be $4.93 \pm 0.05$ , and the pH of Extraction Fluid #2 verified to be $2.88 \pm 0.05$ immediately prior to use?	5.7.2					
Were preservatives not added to the sample prior to extraction? (They should not be.)	6.3					
If precipitation occurred during storage, was the entire sample including precipitate included in the extraction process?	6.4					
Were samples intended for volatile analyses stored at 4°C to minimize volatile loss?	6.5					
Were samples collected using a sampling plan?	6.1					
Were extracts intended for metals analyses acidified with nitric acid to a pH < 2?	6.6					
Were extracts intended for organics analyses stored in containers with no headspace or atmospheric contact?	6.6					
Were extracts stored for acceptable holding times?	6.6					
Were the proportions of solids in samples determined preliminarily to TCLP extractions?	7.1					
Notes/Comments:						

EPA 1311 1992 Page 2 of							
Relevant Aspect of Standards	Method Reference	Υ	N	N/A	Comments		
When percent solids were greater than 0.5% and Volatiles were Involved							
Were particle sizes reduced when particle sizes exceeded 1 cm in narrowest dimension or solids had a surface area greater than 3.1 cm <sup>2</sup> /g?	7.1.3						
Where volatiles were to be determined, were solids reduced to a sieve size of 1 mm and 5.0 grams of the solid phase diluted with 96.5 mL of reagent water and stirred?	7.1.4.1-2						
If the pH of the diluted solid was >5.0, was the pH reduced to <5.0 with 3.5 mL 1N HCL and the sample covered with a watch glass and digested at 50°C for 10 minutes?	7.1.4.3						
If the pH was <5.0 was Extraction Fluid #1 used, and, if the pH was >5.0, even after the above digestion, was Extraction Fluid #2 used?	7.1.4.3-4						
If samples were determined to be 100% solid, were they not subjected to the above three checklist items?	7.1.5						
When percent solids were greater than 0.5% and Vo	platiles were no	t In	volv	ed	l		
Where samples were liquid or multiphase, were the liquid and solid separated by filtration?	7.2.1						
Were particle sizes of solid phases reduced when particle sizes exceeded 1 cm or solids had a surface area greater than 3.1 cm <sup>2</sup> /g?	7.2.9						
Was the amount of extraction fluid to be added to the extractor vessel determined by ((20 x percent solids x weight of waste filtered)/100)	7.2.11						
Was extraction device rotated at $30 \pm 2$ rpm for $18 \pm 2$ hours at a room temperature of $23 \pm 2^{\circ}$ C?	7.2.12						
Were materials in the extractor filtered through a glass fiber filter following extraction?	7.2.12						
Were extraction device product filtrates recombined with liquid filtrates from previous steps if necessary	7.2.13						
Were the pHs of extracts recorded and immediately aliquoted and preserved if necessary and stored at proper temperatures?	7.2.14						
Notes/Comments:				1			

7.3 7.3 7.3 7.3	Y	N	N/A	Comments	
7.3					
7.3					
7.3					
7.3.12.3					
8.1					
8.2					
8.2.1					
8.5					
	8.2.1	8.2.1	8.2.1	8.2.1	8.2.1